



A correlation between superheated liquid fragility and supercooled liquid fragility in La- and Sm-based glass-forming alloys

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ABSTRACT

It is deduced that the parameter of supercooled liquid fragility, m , is proportional to the product of reduced glass transition temperature, T_g/T_L (T_g is the glass transition temperature, T_L the liquidus temperature) and reciprocal of the parameter of superheated liquid fragility, $1/M$, based on the definitions of the two kinds of liquid fragility and the previously studied results. The experimental result in La- and Sm-based glass-forming alloys is consistent with the deduced result. The correlation relates superheated liquid fragility with supercooled liquid fragility.

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1. Introduction

The concept of supercooled liquid fragility can date back to the mid of 1980s. Since then, it has attracted much attention from scientific and technological view points and thus has been studied both theoretically and experimentally [1–3]. Concerning the fragility, there are various quantitative definitions [4–7]. The most commonly used one was introduced by Angell, referring to deviations from Arrhenius temperature dependence of relaxation time or viscosity [4]. According to the definition of liquid fragility, Angell suggested that glass-forming liquids can be classified into three categories, i.e., strong glass formers, fragile glass formers, and intermediate glass formers. Strong liquids exhibit nearly Arrhenius temperature dependence of those dynamic properties. Fragile liquids generally display Vogel–Fulcher–Tamman behavior [8–10]. The slope of the temperature dependence of viscosity at T_g (glass transition temperature) is defined as a fragility parameter, m , to evaluate the fragility of different liquids [11],

$$m = \left. \frac{d \log_{10} \tau(T)}{d(T_g/T)} \right|_{T=T_g} = \left. \frac{d \log_{10} \eta}{d(T_g/T)} \right|_{T=T_g} \quad (1)$$

The values of fragility parameter typically range between $m = 16$ for strong glass formers and $m = 200$ for fragile glass formers. Intriguingly, it has been found that m has a negative correlation with the glass-forming ability for some glass formers [12] and recently researchers also found there is a close relationship

between m and the glass properties such as vibration, Poisson's ratio [13–15]. Novikov et al. presumed that the steepness of the temperature variations of the relaxation time or viscosity near T_g is determined by the high-temperature behavior of the liquid [14].

It seems that supercooled liquid fragility is becoming one of the crucial subjects in the field of metallic materials. Nevertheless, it is usually difficult to determine m for the glass formers with the low GFA since the glassy solids should be prepared beforehand and the sample size must be several millimeters at least. In view of the relatively easy measurement on viscosity of superheated liquid, Bian et al. proposed the concept of superheated liquid fragility [16]. They also found that the superheated liquid fragility can predict the GFA of the Al-based alloys. Based on the thermodynamic refinement, Meng et al. proposed the refined superheated fragility parameter ($M^* = M/\varepsilon$) for evaluating the GFA of BMGs [17]. It can be understood that not only the supercooled liquid fragility, the superheated liquid fragility are also closely related to the GFA of BMGs. Since they both closely correlate with GFA, there should be a relationship existing between the supercooled liquid fragility and the superheated liquid fragility. The correlation between them will be a further insight into the nature of the glass formation.

In this paper, we will deduce the correlation between m with M based on the definition of the two kinds of fragility and some previously studied results. Then the experimental results of Sm- and La- based glass-forming alloys will also be used to demonstrate the relationship between them.

2. Experimental procedure

The LaAlNi(Cu) and SmAlCo(Ni) alloy ingots with nominal compositions were made by alloying high-purity elements (99.9 wt% at least) in an arc furnace under

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argon atmosphere. To ensure homogeneity of the samples, the ingots were remelted at least 5 times. Finally, cylindrical rods of 26 mm × 40 mm for viscosity measurement were prepared by suction casting method in an arc furnace under argon atmosphere. The temperature-dependent viscosities in superheated liquid state were measured using an oscillating vessel viscometer (RHEOTRONI C VIII) under a high vacuum atmosphere. The detailed measurement process for the superheated liquid can see the descriptions of Meng et al. [18]. In order to obtain supercooled liquid fragility, the glassy cylindrical rods with a diameter of 2 mm and a length of 60 mm were prepared by suck-casting into a Cu mold. The structure of the transverse cross sections of as-cast alloys was ascertained using X-ray diffraction (XRD, Cu K α radiation). A differential scanning calorimeter (Netzsch DSC 404C) was used to study the thermodynamic properties at different heating rates.

3. Results and discussion

As we know, all viscosity or relaxation time curves should intersect at two points on the fragility plots: one is T_g , where the viscosity is 10^{13} Poise or the relaxation time is 10^2 s; the other is the very high temperature, $T_g/T \rightarrow 0$, where all liquids have $\eta = \eta_0 = 10^{-4}$ Poise or the relaxation time, $\tau = \tau_0$, is 10^{-14} s. [14]. It indicates that, in fragility plot, a liquid with a steeper slope of $\log \eta$ near T_g at the supercooled state should have a smaller slope of $\log \eta$ at high temperatures in the superheated liquid state. It suggests that the viscosity properties of supercooled liquids can be got by studying the viscosity properties of superheated liquids. Novikov et al. has found that a good linear correlation exists between fragility and T_g/E based on the experimental data of some glass formers [14], as the following

$$m \propto \frac{T_g}{E}, \quad (2)$$

where T_g is glass transition temperature, E is active energy. The parameter of fragility, m , can be got by the thermodynamics method, by which the values of liquid fragility parameter are consistent with those obtained by kinetic method [6]. Previous studies demonstrated that the Q (the heating rate) dependent T_g describes the fragility equally well as the temperature dependence of viscosity [19]. Therefore, the dependence of T_g on Q can also be described by a VFT relation,

$$Q^{-1} = A \exp\left(\frac{DT_0}{T - T_0}\right) \quad (3)$$

$$m = \frac{DT_0 T_g}{(T_g - T_0)^2 \ln 10} \quad (4)$$

where A is a constant, T_0 is the VFT temperature, and D is the strength parameter.

The fragility parameter of superheated liquid was defined as [16],

$$M = \left| \frac{\partial \eta(T)/\eta(T_L)}{\partial T/T_L} \right| = \left| \frac{\partial \eta_r}{\partial T_r} \right| \quad (5)$$

which can be simplified as,

$$M = \frac{E}{RT_L} \quad (6)$$

where T_L is the liquidus temperature, E is active energy and R the gas constant. Actually, M , is characterized by the temperature dependence of the viscosity scaled by the viscosity at the liquidus temperature. Since it reflects the bonding nature of atoms in liquids, viscosity exhibits a close correlation with the liquid phase stability. The low value of the M , means a stronger interaction among structural entities in the superheated liquids and slow structure rearrangement towards the liquidus temperature, whereas a liquid with high value of M would change rapidly in structure upon solidification.

Table 1

The fragility parameters of supercooled liquid and superheated liquid, m and M , respectively, the liquidus temperature, T_L , the glass transition temperature, T_g , the reduced glass transition temperature, T_g/T_L , and $(T_g/T_L) \times (1/M)$ of La- and Sm-based glass-forming alloys.

	m	M	T_L	T_g	T_g/T_L	$(T_g/T_L) \times (1/M)$
La ₅₅ Al ₂₅ Ni ₂₀	33	1.7747	941	491	0.521	0.2936
La ₅₅ Al ₂₅ Ni ₁₅ Cu ₅	36	1.6428	900	474	0.527	0.3208
La ₅₅ Al ₂₅ Ni ₅ Cu ₁₅	42	1.4509	878	459.1	0.523	0.3605
Sm ₅₀ Al ₃₀ Co ₂₀	33	1.9588	960	586	0.610	0.3114
Sm ₅₅ Al ₂₅ Ni ₁₀ Co ₁₀	36	1.8260	898	553	0.616	0.3373
Sm ₅₅ Al ₂₅ Cu ₁₀ Co ₁₀	28	2.1005	986	534	0.54	0.2861

Substituting Eq. (6) into Eq. (2), we can get the correlation between m and M ,

$$m \propto \left(\frac{T_g}{T_L}\right) \left(\frac{1}{M}\right) \quad (7)$$

Eq. (7) means that the parameter of supercooled liquid fragility is proportional to the product of reduced glass transition temperature, T_g/T_L , and reciprocal of the parameter of superheated liquid fragility, $1/M$, in bulk glass-forming alloys. Actually, the reduced glass transition temperature is one of the criterions for estimating GFA of glass-forming alloys proposed by Turnbull based on the assumption that the nucleation frequency of a melt scales as $1/\eta$ (η , viscosity of the liquid) [20]. Accordingly, the equation indicates the relationship between the supercooled liquid fragility and superheated liquid fragility and also suggests that both of them are indeed closely related to GFA of glass-forming alloys.

LaAlNi(Cu) and SmAlCo(Ni) alloys were well chosen to investigate the correlation between m and M . The parameters of supercooled liquid for La-based alloys, determined by the thermodynamic method, were cited from the reference [21,22]. The values of m for Sm-based alloys were obtained by the Eqs. (3) and (4) based on DSC experiments at different heating rates, as shown in Table 1. Fig. 1 displays DSC curves of the Sm₅₅Al₂₅Cu₁₀Co₁₀ BMG with different heating rates. Figs. 2 and 3 are the superheated liquid fragility plots of La- and Sm-based glass formers, respectively. In the figures, R^2 is the square of correlation coefficient, reflecting the error of the fitting curve. The parameters of superheated liquid fragility determined by Eq. (6) are also list in Table 1.

Based on the data listed in Table 1, the plots of the correlation between m and $(T_g/T_L) \times (1/M)$ are shown in Fig. 4. R is the correlation coefficient, reflecting the error of the fitting line. It can be obviously observed that a good linear correlation exists

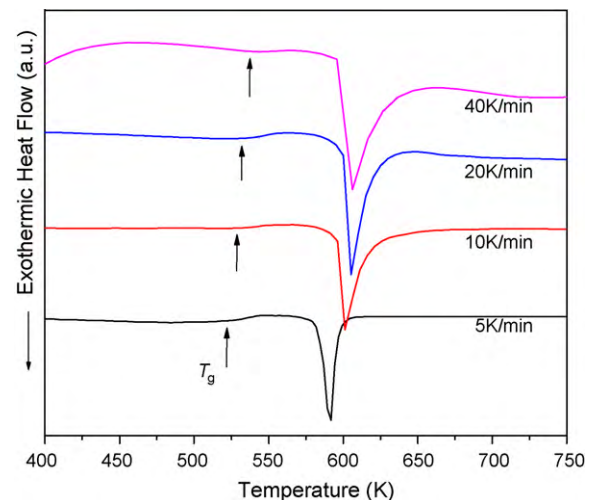


Fig. 1. DSC curves of the Sm₅₅Al₂₅Cu₁₀Co₁₀ BMG with different heating rates.

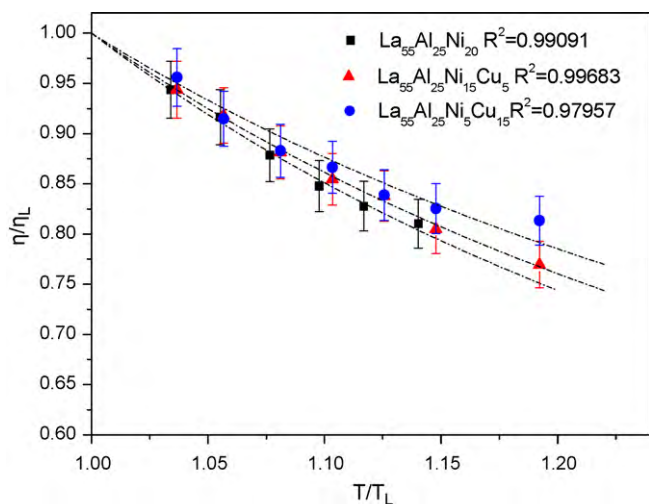


Fig. 2. The scaled experimental viscosity data (as single points) and fitting curves (as continuous curves) for La-based system alloys.

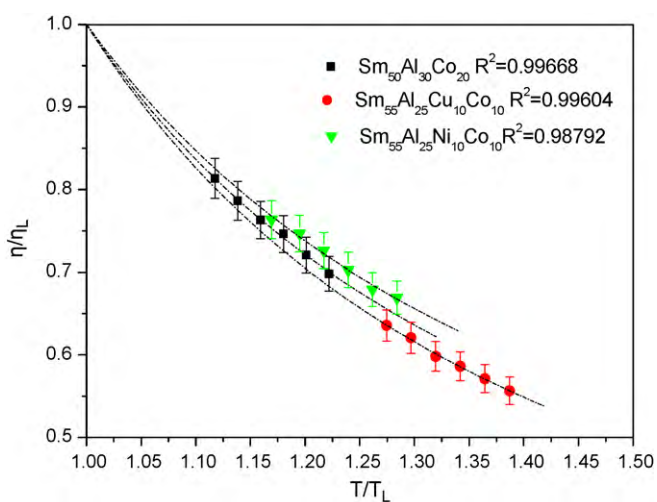


Fig. 3. The scaled experimental viscosity data (as single points) and fitting curves (as continuous curves) for Sm-based system alloys.

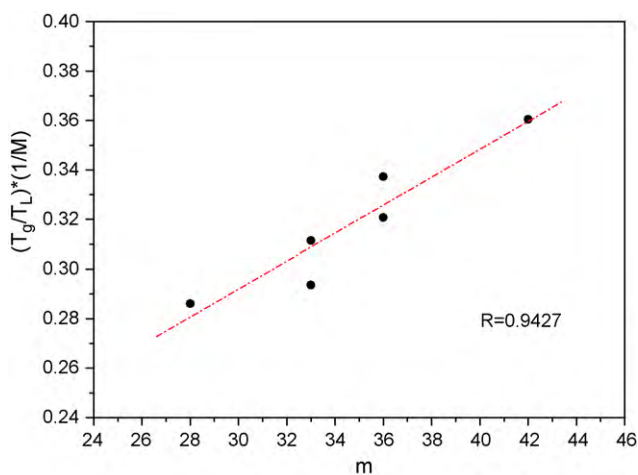


Fig. 4. Correlation between the parameter of supercooled liquid fragility, m , and $(T_g/T_L) \times (1/M)$ of the La- and Sm-based glass-forming alloys.

between them for La- and Sm-based glass-forming alloys. The experimental result in La- and Sm-based alloys is consistent with the result deduced from the definitions of the two kinds of liquid fragility.

The correlation between the fragilities of supercooled liquid and superheated liquid is of significance to estimate the fragility of supercooled liquid and even the GFA of an alloy by the investigation on high-temperature behavior of the normal liquid. It has been noticed that although great progress has been made in the work concerning the fragility of supercooled liquid, some problems restricting the investigation on the fragility of supercooled liquid still remain. Especially for the liquid with an unstable supercooled liquid state, the glassy solids should be prepared beforehand. Hence, for the alloys with low glass-forming ability, it is nearly impossible to obtain the fragility of supercooled liquid. In this case, the fragility can be estimated via the behaviors of a high-temperature liquid and its liquidus temperature without any reference to the glass transition or supercooled regime of the liquid state.

4. Conclusion

The correlation between the fragilities of supercooled liquid and superheated liquid was investigated based on the definitions of the two kinds of liquid fragility and the previously studied results. The parameter of supercooled liquid fragility, m , was found to be proportional to the product of the reduced glass transition temperature, T_g/T_L and reciprocal of the parameter of superheated liquid fragility, $1/M$, in glass-forming alloys. The experimental result in La- and Sm-based glass-forming alloys was consistent with the deduced result. The relationship between the fragilities of supercooled liquid and superheated liquid is a further insight into the nature of the glass formation.

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References

- [1] C.A. Angell, *Science* 267 (1995) 1924–1935.
- [2] F. Sciortino, P. Tartaglia, E. Zaccarelli, *Phys. Rev. Lett.* 95 (2005), 15802–1–3.
- [3] T. Scopigno, G. Ruocco, F. Sette, G. Monaco, *Science* 302 (2003) 849–852.
- [4] C.A. Angell, *J. Non-Cryst. Solids* 73 (1985) 1–17.
- [5] L.M. Martinez, C.A. Angell, *Nature* 410 (2001) 663–667.
- [6] K. Ito, C.T. Moynihan, C.A. Angell, *Nature* 398 (1999) 492–495.
- [7] D.H. Huang, B. Gregory, Mckenna, *J. Chem. Phys.* 114 (2001) 5621–5630.
- [8] H. Vogel, *Phys. Z* 22 (1921) 645–646.
- [9] G.S. Fulcher, *J. Am. Ceram. Soc.* 8 (1923) 339–355.
- [10] G. Tamman, W. Hesse, *Z. Anorg. Allg. Chem.* 156 (1926) 245–257.
- [11] D.J. Plazek, K.L. Ngai, *Macromolecules* 24 (1991) 1222–1224.
- [12] R. Busch, A. Masuhr, E. Bakke, *Mater. Res. Soc. Symp. Proc.* 455 (1997) 369–374.
- [13] V.N. Novikov, A.P. Sokolov, *Nature London* 431 (2004) 961–963.
- [14] V.N. Novikov, Y. Ding, A.P. Sokolov, *Phys. Rev. E* 71 (2005), 061501–1–12.
- [15] W.H. Wang, *J. Appl. Phys.* 99 (2006), 093506–1–9.
- [16] X.F. Bian, B.A. Sun, L.N. Hu, Y.B. Jia, *Phys. Lett. A* 335 (2005) 61–67.
- [17] Q.G. Meng, S.G. Zhang, M.X. Xia, J.G. Li, J.K. Zhou, *Appl. Phys. Lett.* 90 (2007), 031910–1–3.
- [18] Q.G. Meng, J.K. Zhou, H.X. Zheng, *Scr. Mater.* 54 (2006) 777–781.
- [19] R. Busch, E. Bakke, W.L. Johnson, *Acta Mater.* 46 (1998) 4725–4732.
- [20] D. Turnbull, *Contemp. Phys.* 10 (1969) 473–488.
- [21] Z.P. Lu, X. Hua, Y. Lia, S.C. Ng, *Mater. Sci. Eng. A* 304–306 (2001) 679–684.
- [22] Z.P. Lu, C.T. Liu, *Acta Mater.* 50 (2002) 3501–3512.